(acetoxy carbonyl), 1385, 875 (terminal methylene). NMR, δ 4.30 (2H, d, cis olefinic protons at C-6 and C-7, 4.21 (2H, s, terminate methylene protons at C-24), 4.15–3.75 (1H, br, H-3), 1.69 (3H, s, acetoxy methyl), 1.1–1.55 (22H, complex, m, ring methylene protons, side chain methylene protons and methine protons at C-9, C-13, C-17 and C-25), 0.70 (6H, s, gem dimethyl at C-26 and C-27), 0.83 (3H, s, Me-18), 0.9 (3H, s, deshielded tertiary methyl protons at C₂₁), 0.45 (3H, s, Me-19). (Found: C, 79.28, H, 10.24. C₃₀H₄₆O₃ requires: C, 79.25; H, 10.20%.)

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REFERENCES

- Takemoto, T., Ogawa, S., Mishimoto, N., Yen, K. Y., Abe, K., Sato, T. and Takashi, M. (1967) Yakugaku Zasshi 87, 1521.
- Takemoto, T., Hikino, Y. and Mishimoto, N. (1967) Yakugaku Zasshi 87, 325.
- 3. Takemoto, T., Nomoto, N. and Himino, H. (1968) Tetrahedron Letters. 4953.
- 4. Galbraith, M. N., Horn, D. H. S., Middleton, E. J. and Hackney, R. J. (1968) Chem. Commum. 466.
- Bhacca, S. N. and Williams, D. H. (1964) in Application of NMR Spectroscopy in Organic Chemistry. Holden-Day, San Francisco.
- 6. Cohen, C. F., Louloudes, S. J. and Thompson, M. J. (1967) Steroids 9, 591.

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A DAMMARANE TRITERPENE FROM COMMELINA UNDULATA*

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Key Word Index—Commelina undulata; Commelinaceae; dammarane triterpene; dammar-12, 25-dien-3β-acetate.

Abstract—The chemical investigation of Commelina undulata afforded, in addition to 2-heneicosanone, n-octacosanol, sitosterol and sitosterol- β -D-glucoside, a new triterpene characterized as dammar-12,25-dien- 3β -acetate.

INTRODUCTION

In the course of our search for anti-cancer constituents in Indian plants, we have examined Commelina undulata which possessed anti-cancer activity against lymphoid leukaemia in mice (PS 388) in the screening programme of NIH. A new triterpene (1) of the dammarane series was isolated together with 2-heneicosanone, n-octacosanol, sitosterol and sitosterol- β -D-glucoside.

RESULTS AND DISCUSSION

Systematic fractionations of an alcoholic extract of the plant was carried out with hexane, benzene, ethyl acetate and *n*-butanol. The crystalline compound 1 was isolated by CC from the hexane fraction over Si gel.

The IR spectrum of compound 1 had absorption bands at 1740 (C = O); 1384, 1370 (gem methyls); 1245 (C-O-C), 1660, 1640, 985 and 885 cm^{-1} (C=CH and

C=CH₂). The UV spectrum had λ_{max} at 240 nm (ϵ = 523). The compound exhibited a positive Liebermann-Burchardt test and gave a yellow colouration with tetranitromethane. The 'H NMR spectrum showed five tertiary methyl signals at δ 0.73, 0.8, 0.82 (6H) and 0.88 and a signal at 0.95 (3H, d, J = 7 Hz), assigned to Me-20[1]. The methylene and methine protons were in the range $\delta 1.55-1.08$. A vinylic methyl appeared at 1.6 (3H, s) and an acetoxyl methyl singlet was at δ 1.95. A multiplet was centered at 2.25 (4H, H-11 and H-24) and the signals at 4.6 and 4.48 (d, J = 2 Hz) were assigned to the C-26 methylene protons. The signal at 4.36 (1H, dd, $J_{ax,ax}$ = 10 Hz, $J_{ax, eq} = 6.5$ Hz,) was assigned to the C-3 α proton and an olefinic proton was observed at 5.1 (1H. m. H-12). The mass spectrum of this compound had the M⁺ at m/z 468 and other ions at m/z 453 O

 $[M-Me]^+$, 425 $[M-C-Me]^+$ and 408 $[M-MeCO_2 H]^+$. The presence of side chain and C-12 double bonds in the molecule was evident from the ion peak at m/z

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357 $[M - C_8H_{15}]^+$ and the prominent fragment at m/z 218 (3) formed through retro Diels-Alder fission of ring C.

From the above spectral data it is evident that the compound belongs to the dammarane series [1, 2] with an acetoxyl group at C-3 and two double bonds, one in the C-ring and the other in the side chain. The presence of the side chain double bond was further confirmed through catalytic hydrogenation of compound 1 which gave a dihydro product, dammarenyl acetate, mp 197° (MeOH), $C_{32}H_{54}O_2$ (M⁺470). The signal at 0.85 (6H, d, J = 7 Hz) and the absence of a vinylic methyl and methylene proton signals in the ¹H NMR spectrum of the dihydo product established the presence of a terminal isopropyl group in the molecule.

Alkaline hydrolysis of compound 1 yielded the deacetylated product 2, mp 182–184° (MeOH), $[\alpha]_D$ + 116° (CHCl₃; c 0.82), $C_{30}H_{50}O(M^+426)$. Its IR spectrum had a band at 3400 cm⁻¹ (-OH). The ¹H NMR spectrum had signals at δ 3.36 (1H, dd, $J_{ax,ax}$ = 10 Hz, $J_{ax,eq}$ = 6 Hz) which were assigned to the secondary carbinol proton at C-3, which was deshielded by δ 1.0 on acetylation.

The major product isolated after selenium dioxide oxidation of dammarenyl acetate was the heteroannular diene, 4 [3], mp $163-164^{\circ}$ (MeOH), (M⁺468). The UV spectrum of compound 4 had three λ_{max} at 242 ($\epsilon = 6598$) 250 ($\epsilon = 32\,090$) and 260 nm [4–6] and the IR spectrum had a C-H deformation band at 975 cm⁻¹, characteristic for a trans-diene. The ¹H NMR spectrum of compound 4 displayed one of the ethylenic protons at $\delta 5.4$ (dd, J = 10, 1.5 Hz, H-12) and the other at 6.25 (dd, J = 10, 3.2 Hz, H-11).

Column chromatography of the benzene fraction over Si gel yielded a minor compound 2 $C_{30}H_{50}O$ (M⁺426), mp 182–83° (MeOH), which was found to be identical in all respects (mmp, IR, NMR, MS and co-TLC) with the compound obtained through the alkali treatment of dammar-12, 25-dien-3 β -acetate (1).

EXPERIMENTAL

Mps are uncorr.

Isolation of dammaradienyl acetate (1). The air-dried powdered herb (10 kg) was extracted with EtOH. The

residue obtained was fractionated successively with C_6H_{14} (5 × 2 l.), C_6H_6 (3 × 1 l.), EtOAc (3 × 1 l.) and n-BuOH (2 × 1 l.). The C_6H_{14} extract (238 g) was concd and chromatographed over Si gel. Elution with C_6H_{14} - C_6H_6 (1:1) yielded dammaradienyl acetate (1) crystallized from MeOH (0.5 g) mp 177–178°; $[\alpha]_D$ + 113° (pyridine c 0.7). (Found: C, 81.55; H, 10.69; Calc. for $C_{32}H_{52}O_2$, C, 82.05; H, 11.11%.)

Dammarenyl acetate. A soln of 1 (50 mg) in EtOAc was hydrogenated with PtO₂ for 3 hr. The product was crystallized from MeOH, mp 197°; [M]⁺ at m/z 470 (C₃₂H₅₄O₂); IR $\nu_{\rm max}^{\rm KBr}$ cm⁻¹: 2925, 1735, 1450, 1380, 1365, 1245, 1100, 1025 and 800; ¹H NMR (90 MH₃, CDCl₃): δ 0.75 (3H, s), 0.78 (3H, s), 0.82 (6H, s), 0.84 (3H, s), 0.85 (6H, d, d) = 7 Hz), 0.99 (3H, d), d

J = 7 Hz), 1.1–1.8 (23H, m), 0.95 (3H, s, -C-Me), 2.64 (2H, m), 4.4 (1H, dd, J = 10, 6 Hz) and 5.04 (1H, m).

Selenium dioxide oxidation. A soln of the dammarenyl acetate (30 mg) in AcOH (5 ml) and H₂O (2 ml) was refluxed with freshly sublimed SeO₂ (60 mg) for 10 hr. After adding NaOAc it was again refluxed for another 2 hr. The reaction mixture was poured into 50 ml H₂O and extracted with CHCl₃. The CHCl₃ layer was washed with NaHCO₃ and then with H₂O and finally dried over Na₂SO₄. The products were separated by prep. TLC on Si gel. The major product was a heteroannular diene (4), mp 163° (MeOH) IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 2900, 1730, 1700 (sh), 1460, 1380, 1365, 1250, 1025, 980 and 800; UV λ_{max}^{MeOH} nm: 242 (ϵ = 6598), 250 $(\epsilon = 32\,090)$ and 260 nm; ¹H NMR (CDCl₃): $\delta 0.7$ (3H, s), 0.78 (3H, s), 0.8 (6H, s), 0.85 (3H, s), 0.85 (6H, d, J = 7 Hz), 0.95(3H, d, J = 7 Hz), 1.0-1.85 (20H, m), 1.98 (3H, s), 2.1-2.7(4H, m), 4.4 (1H, dd, J = 10, 6 Hz), 5.4 (dd, J = 10, 1.5 Hz)H-12) and 6.25 (dd, J = 10, 3.2 Hz, H-11); MS, m/z: 468 [M]⁺, 451, 410, 406, 271, 229, 218 (100%), 203, 189, 133, 135, 121, 119, 113, 111 and 107.

Alkaline hydrolysis of. A soln of 1 in 5% methanolic KOH was refluxed for 4 hr to give dammaradienol (2) mp 182–83° (MeOH), $[\alpha]_D + 116^\circ$ (CHCl₃; c0.82), IR, $\nu_{\rm max}^{\rm KBr}$ cm⁻¹: 3350, 1660 (sh), 1640, 1460, 1380, 1360, 1180, 1140, 1100, 1030, 990, 880 and 760 ¹H NMR(CDCl₃): 0.7, 0.72, 0.75, 0.88, 0.9 (each singlet equivalent to 3H), 0.95 (3H, d, J = 6 Hz) 1.3–1.5 (1H, m, OH), 3.35 (1H, dd, J = 10, 6 Hz), 4.48, 4.6 (each equivalent to 1H, both d, J = 2 Hz, H-26H) and 5.2 (1H, m); MS, m/z: 426 [M]⁺, 411, 408 [M – H₂O]⁺, 393, 315, 313, 218 (100%), 203, 189, 175, 161, 149, 147, 135, 123, 121, 111, 109, 107 and 97.

REFERENCES

- Barnes, C. S., Galbraith, M. N., Ritchie, E. and Taylor, W. C. (1965) Aust. J. Chem. 18, 1411.
- Talpatra, S. K., Bhar, D. S. and Talapatra, B. (1974) Aust. J. Chem. 27, 1137.
- Budziarek, R., Manson, W. and Spring, F. S. (1951) J. Chem. Soc. 3336.
- 4. Scott, A. I. (1964) Interpretation of the Ultraviolet Spectra of Natural Products Vol. 7, p. 51.
- 5. Fischer, G. F. and Seiler, N. (1961) Ann. Chem. 644, 162.
- Arigoni, von D., Jeger, O. and Ruzicka, L. (1955) Helv. Chim. Acta 38, 222.